

Reading Tip

As you read, check the predictions you made to see if they were correct.

Target Your Reading

Use this to focus on the main ideas as you read the chapter.

- 1 Before you read** the chapter, respond to the statements below on your worksheet or on a numbered sheet of paper.
 - Write an **A** if you **agree** with the statement.
 - Write a **D** if you **disagree** with the statement.
- 2 After you read** the chapter, look back to this page to see if you've changed your mind about any of the statements.
 - If any of your answers changed, explain why.
 - Change any false statements into true statements.
 - Use your revised statements as a study guide.

Before You Read A or D	Statement	After You Read A or D
	1 Burning is an example of a chemical change.	
	2 A chemical equation only tells the names of reactants and products.	
	3 When a substance burns, atoms disappear and new atoms are created.	
	4 When balancing a chemical equation, it's okay to change the subscripts of a chemical formula.	
	5 Some reactions release energy and some adsorb energy.	
	6 During chemical reactions, bonds in the reactants break and new bonds form.	
	7 Reactions that release energy do not need any energy to start the reaction.	
	8 Increasing temperature will speed up most chemical reactions.	